

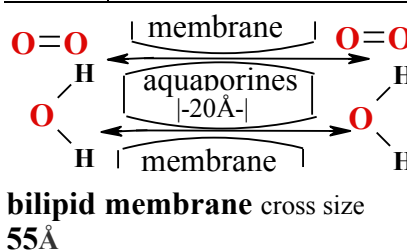
The **Water channels**, allow the **passive** but **selective** movement of **Water** and **O₂, NO, CO** across **cell wall** and **subcellular membranes** like as **mitochondria**, **endoplasmic reticulum**, **peroxisomes**, **Golgi**, **lysosomes**.... . **Aquaporins** have been classified into **two** sub-families:

I) strict **Aquaporins** that only allow the passage of **Water**, **O₂**, **NO**, **CO** and II) the **less selective aquaglyceroporins** that transport **Water** and other neutral **solutes**, such as **Glycerol**, **CO₂** or **urea**.

Recently, the identification and characterization of a number of archaeal and bacterial **Aquaporins** suggested the existence of a **third** sub-family; one that is neither a strict **Aquaporin** nor an **aquaglyceroporin**. The function and phylogeny of this **third** family is still a matter of debate.

Water channels H₂O common O₂, NO, CO: an overview

AQP0	+ Cl⁻, NO₃⁻ eye-lens cells; thin junctions between fiber cells AQP0 with a measured Water permeability 15-fold lower than that of AQP1 at pH 6.5 ; AQP0 is reduced a further three fold at pH 7.5 AQP0 induce a gating effect close conformations of extracellular loop A Met176, His40 AQP0 becomes more constrained near the conserved Ar/R constriction site
AQP1-	+ Cl⁻, NO₃⁻ , Aquaglyceroporins: red blood cell (RBC), apical & basolateral membranes of epithelial brain cell, rodent brain cell AQP1-null humans kidney proximal-tubule water reabsorption gastrointestinal tract Water absorption in the teleost intestine the ovary and in the oocyte ; salivary gland ;
AQP2	urinary bladder granular kidney cells & subcellular vasopressin regulated urine concentration (25% of the blood filtrate) trans located from the cytoplasmic pool to the apical plasma membrane of the granular cells of the pelvic patch and urinary bladder
AQP3	+ Aqua glyceroporins, urea: gastrointestinal tract Water absorption; rodent brain cell astrocyte end-feet Water enters in the principal cell through AQP2 and exits through located in the basolateral membranes trachea basal AQP3 & ciliated columnar AQP4 cells
AQP4	Rodent-brain; basolateral membrane of ciliated columnar cells alveolar epithelium; salivary gland
AQP5	stomach, duodenum, pancreas, airways, lungs, salivary gland, sweat glands, eyes, lacrimal glands, and the inner ear tears & pulmonary sub mucosal glands secretions apical membrane & rodent brain cells
AQP6	+ Cl⁻, NO₃⁻ multi permeable channel; lens cells; may play a role in the body acid-base homeostasis in the intracellular vesicles of acid-secreting intercalated cells of the RCD colocalized with the H ⁺ -ATPase be Hg ²⁺ -inhibit able Water channel function is activated by Hg ²⁺ and low pH
AQP7	+ Aquaglyceroporins, urea; kidney proximal tubule epithelium cell glycerol reabsorption ; together with AQP1 in the brush border in the concentration of urine taking place in the proximal nephron 75% of the blood filtrate which is 150–180 L per day
AQP8	NH ₄ ⁺ ; lens & kidney intracellular proximal tubule & small intestine absorptive: epithelium cell in the concentration of urine taking place in the proximal nephron also in mitochondria 75% of the blood filtrate which is 150–180 L per day & rodent brain cell
AQP9	+ Aquaglyceroporins, urea purines, pyrimidines & monocarboxylates, arsenite ; apical membrane of brain & small intestine absorptive epithelial & rodent brain & glial cells
AQP10	+ Aquaglyceroporins, urea ; small intestine absorptive epithelial cells
AQP11	“super aquaporins” or sub cellular; kidney cytoplasm of the proximal tubule & rodent brain cells
AQP12	“super aquaporins” or sub cellular



channel proteins (WCPSs) are **trans membrane proteins** that have a specific three-dimensional structure with a **pore** the **SF radius 1.1 Å** is close average to **radius of water H-O-H** longitudinal **1.4 Å** and **0.55 Å** bent size of dipole. It can be permeated by **Water & O₂, NO, CO** molecules as solutes. **Aquaporins** are large families (over **450 members**) that are present in **all kingdoms of life**. **Water** permeability, allowing permeation of **3 × 10⁹ water molecules per monomer per second** AQP1 and other, which strictly prevents the conduction of protons **H⁺**.

Serine, Tyrosine, Threonine membrane trafficking of AQP1, AQP2, AQP5, and AQP8, and the gating of AQP4. Phosphorylation to trigger the me Cation conductance has been induced in AQP1 by activation of cyclic GMP-dependent pathways and was blocked by Hg²⁺

Red blood cells against colligative osmomolar concentration in **water** solutions

Water and oxygen osmosis against osmo molar concentration gradient crosses cell membranes

Osmosis is organised for H_2O and O_2 movement against concentration gradients-difference of colligative properties $\Delta C_{\text{osm}} = i\Delta C_M$ through an **Aquaporins** across cell **membranes** to form the osmotic pressure:

$$\pi = i\Delta C_M RT \text{ (kPa)},$$

where $R=8,3144 \text{ J}/(\text{mol}\cdot\text{K})$ universal gas constant,

T temperature in Kelvin's degree (K) $T = t + 273.15$ (if $t=37^\circ$ than $T=37 + 273.15=310.15 \text{ K}$).

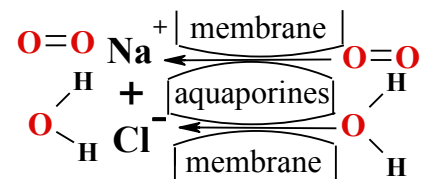
Note: Transfer **water** and oxygen molecules through membrane aquaporin tunnel in erythrocytes with rate $3 \cdot 10^9 \text{ sec}^{-1}$ in both directions transfer 3000 oxygen molecules in one second.

Mechanism of osmosis through membrane aquaporins drive colligative concentration gradient

$\text{Na}^+\text{Cl}^- \Rightarrow \text{Na}^+ + \text{Cl}^-$ $m=2$ electrolyte dissociation $\alpha=1$ the concentration gradient doubled as i is 2

$i=1+\alpha(m-1)=1+1(2-1)=2$; $i\Delta C_M = 2\Delta C_M = \Delta C_{\text{osm}}$ and pressure on membrane is $\pi = 2\Delta C_M RT = \Delta C_{\text{osm}} RT$.

Press \Rightarrow on membrane to right.



Water H_2O , O_2 oxygen flow left side against the concentration gradient

from 0 to $C_{\text{osm}}=0.305 \text{ M}$ because Na^+Cl^- ions make

osmo molar concentration left side $C_{\text{left}} - C_{\text{right}} = C_{\text{osm}} - 0 = C_{\text{osm}} = iC_M$

and close H_2O , O_2 flow to right side.

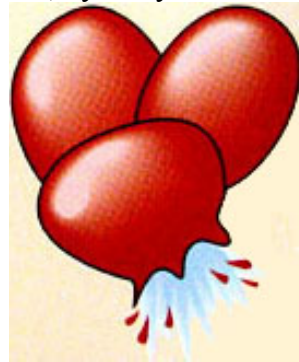
$$C_{\text{blood}} = C_{\text{osm}} = i_1 C_1 + i_2 C_2 + i_3 C_3 + \dots = \sum i_k C_k = 0,305 \text{ M}$$

Human erythrocytes red blood cells with osmo molar concentration 0.305 M of all solutes sum $\sum i_k C_k$:

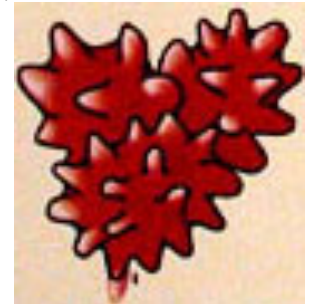
C_{blood} glucose, salts, hydroxonium H_3O^+ , hydroxyl OH^- ions, amino acids, proteins, bicarbonate etc.



Isotonic medium
 $C_{\text{blood}} = 0.305 \text{ M}$



Hypotonic medium
distilled water 0 M
or at least



Hypertonic solution
 $C_{\text{Hyperton}} \geq 0,4 \text{ M}$.

osmo molar concentration $C_{\text{Hypoton}} \leq 0,2 \text{ M}$.

Hypotonic **water** medium the flow is greater towards the cell against the concentration gradient-difference $0.305 - 0,2 = 0,105 \text{ M}$ and the cell puffs up until its membrane is broken but content leak in plasma.

Hypertonic salt solutions to apply for purulent wounds, because pumps **water** toxic compounds out and stimulates **blood** circulation.

Osmosis H_2O and O_2 against concentration gradient through alveolar epithelial membrane

A) **Oxygens** O_2 from **AIR** 20.95% $\text{O}_2 \uparrow$ gas assimilation reaction dissolution in water to form $\text{O}_{2\text{aqua}}$ exothermic $\Delta H_r = -55,7 \text{ kJ}/\text{mol}$ and exoergic $\Delta G_r = -27,7 \text{ kJ}/\text{mol}$ as water soluble oxygen :

1) $\text{O}_{2\text{AIR}} + \text{H}_2\text{O} \Leftrightarrow \text{H}_2\text{O} + \text{O}_{2\text{aqua}} + Q + \Delta G$. Penetrate in Human body through aquaporins by concentration gradient from $[\text{O}_2] = 9,768 \cdot 10^{-5} \text{ M}$ to **venous** blood $[\text{O}_{2\text{aqua}}] = 0,426 \cdot 10^{-5} \text{ M}$.

2) $\Delta G_{\text{O}_2} = RT \ln([\text{O}_{2\text{Blood}}]/[\text{O}_{2\text{aqua}}]) = -4,29 \text{ kJ}/\text{mol}$ exoergic entrance human organism;

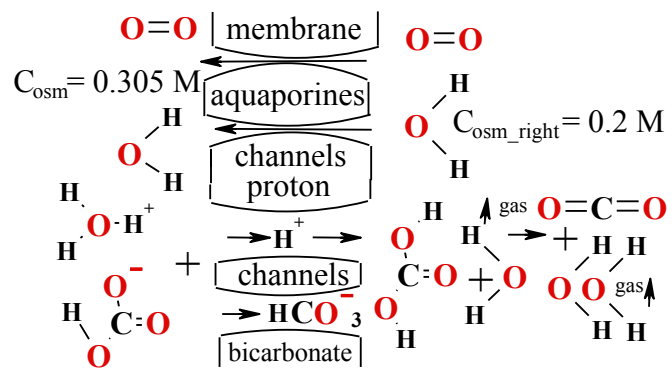
3) $\text{O}_{2\text{aqua}} + \text{H}_2\text{O} \xrightarrow{\text{Aquaporins}} \text{H}_2\text{O} + \text{O}_{2\text{aqua}} + \Delta G$ against concentration gradient 0,305 M / 0,2 M:

$$\Delta G_{\text{H}_2\text{O}} = RT \ln([\text{H}_2\text{O}]_{\text{right}}/[\text{H}_2\text{O}]_{\text{left}}) = -8,3144 \cdot 310,15 \cdot \ln(0,305/0,2) = -1.088 \text{ kJ}/\text{mol}$$

exoergic $\Delta G_{\text{O}_2} = -5,379 \text{ kJ}/\text{mol}$. **Deoxy** hemoglobin **Hb_T** adsorbs 4 $\text{O}_{2\text{aqua}}$ from blood plasma of inspired fresh AIR releases four protons 4H^+ and 4 HCO_3^- stabilizing arterial $[\text{O}_2] = 6 \cdot 10^{-5} \text{ M}$ concentration $4\text{O}_{2\text{aqua}} + (\text{H}^+\text{His63,58})_4 \text{Hb}_T \Leftrightarrow \text{Hb}_R(\text{O}_2)_4 + 4\text{H}^+$.

$$\text{Total exothermic } \Delta H_r = -55,7 \text{ kJ}/\text{mol} \text{ and exoergic } \Delta G_{\text{O}_2} = -27,7 + -4,29 + -1.088 = -33.078 \text{ kJ}/\text{mol}$$

Osmosis is **water** and oxygen flow left side against gradient of concentration 0.2 M to $C_{osm}=0.305$ M because water and oxygen flow to right side closed by made left side osmo molar $C_{left}-C_{right}=C_{osm}-C_{osm_right}=\Delta C_{osm}$ concentration as difference $\Delta C_{osm}=0.105$ M.
 $C_{osm}=\sum i_k C_k=0,305$ M;
 $C_{osm_right}=0,2$ M; $\Delta C_{osm}=0.305-0.2=0.105$ M



Breathe out H_2O , CO_2 in endothermic but exoergic reactions on alveolar epithelial surface

B)

$Q_{aqua} + CO_{2,aqua} + 2H_2O \xleftarrow{CA} H_3O^+ + HCO_3^- \xleftarrow{Membrane} H_3O^+ + HCO_3^- \leftrightarrow H_2O + H_2CO_3 + Q_{gas} \leftrightarrow H_2O + CO_2 \uparrow_{gas} + H_2O$.
 endothermic $\Delta H_r = 9.75$ kJ/mol; athermic $\Delta H_r = 0$ kJ/mol; exothermic $\Delta H_r = -9.76$ kJ/mol; endothermic $\Delta H_r = 20.3$ kJ/mol;
 endoergic $\Delta G_r = 58.4$ kJ/mol; exoergic $\Delta G_r = -22.5-1,96$ kJ/mol; exoergic $\Delta G_r = -58.2$ kJ/mol; exoergic $\Delta G_r = -8,54$ kJ/mol;

B) $Q_{aqua} + CO_{2,aqua} + 2H_2O \xleftarrow{CA} H_3O^+ + HCO_3^- + Q \xleftarrow{Membrane} H_2O + CO_2 \uparrow_{gas} + H_2O \uparrow_{gas}$.

endothermic $\Delta H_r = 9.75$ kJ/mol; endothermic $\Delta H_r = 54,5$ kJ/mol; summary endothermic $\Delta H_r = 64,25$ kJ/mol;

endoergic $\Delta G_r = 58.4$ kJ/mol; exoergic $\Delta G_r = -82,1$ kJ/mol; summary exoergic $\Delta G_r = -23,7$ kJ/mol;

Venous **deoxy Hb_T** shuttle adsorbs four **oxygen** $4O_{2Hb}$ molecules, create $4H^+$, promoting CO_2 breathe out as increase production of H^+ , HCO_3^- $473 \cdot 6 \cdot 10^{-5} M = 0,0284 M = [HCO_3^-] = [H^+]$ amounts shifts equilibrium to right $H^+ + HCO_3^- + Q \leftrightarrow H_2O + CO_2 \uparrow_{gas}$ via membrane channels. So pH=7,36 remains constant, as bicarbonate ion and hydrogen ion produce CO_2 right side.

The epithelial cell surface of **lungs** has the specific building. $S = 950 \text{ nm} \times 950 \text{ nm} = 0.9 \mu\text{m}^2$ is surface area with super thin 0.6 nm **water** layer volume: $0.5415 \cdot 10^{-3} \mu\text{m}^3 = 0.5415 \cdot 10^{-18} \text{ L}$. Created acidity in thin **water** layer volume increases up to pH=5.5 if one proton H^+ crosses the membrane channels reaching the surface. Hydrogen ion concentration is: $[H_3O^+] = 10^{-pH} = 10^{-5.5} \text{ M}$. Respiration in **lungs** Hemoglobin released protons H^+ during oxygen adsorbtion for total amount concentration:

$[O_{2Hb}] = [H_3O^+] = 473 \cdot 6 \cdot 10^{-5} M = 0,0284 M$ forms hydrogen ion concentration gradient:

$[H_3O^+]_{right} / [H_3O^+]_{left} = 10^{-5.5} / 0,0284$, which drives exoergic $\Delta G = -22,5$ kJ/mol proton movement through epithelial cell membrane proton channels: $H_3O^+_{left} \xleftarrow{\text{proton channel}} H_3O^+_{right} + \Delta G$. General process

$H_2O + CO_2 \uparrow_{gas} + H_2O \uparrow_{gas}$ require heat supply endothermic $\Delta H = 54,5$ kJ/mol to drive spontaneous

$\Delta G = -82,0679$ kJ/mol products evaporation $CO_2 \uparrow_{gas}$ and $H_2O \uparrow_{gas}$ keeping moisture H_2O on surface of membrane. Hydrogen ions water acidity shift endothermic $\Delta H_r = +54,5$ kJ/mol and exoergic $\Delta G_r = -82,1$ kJ/mol decomposition $H_3O^+ + HCO_3^-$ breath out to AIR $CO_2 \uparrow_{gas}$ with $H_2O \uparrow_{gas}$:

$H_3O^+ + HCO_3^- + Q \xleftarrow{Membrane} H_2O + CO_2 \uparrow_{gas} + H_2O \uparrow_{gas} + \Delta G_r = -82,1$ kJ/mol. exoergic .

Aquaporins are wide class of **membrane crossing channel** proteins, **which** are **integrated** in all living organisms: **animals, plants, bacteria**. On Cell membranes effecting Physiology, Biochemistry and Health. **Aquaporins** are large families (over 450 members) that are present in all kingdoms of life.